Metric Conversions and Scientific Notation

This workshop reviews the basics of conversions between measurement systems and proper scientific notation that is fundamental to accurate problem solving in STEM fields.

Science is often conducted in a number of different measurement systems depending on discipline or geographic region. In order to effectively communicate scientific findings, scientists often must convert between different systems of measurement using conversion factors. This workshop will give an overview of common conversion factors between the English and metric systems, and will go over the method behind accurate conversions in the biological sciences. Scientific notation is the way of writing and manipulating numbers that are too large or small to be written in the decimal form. We will discuss normalized notation or exponential notation (e.g., \(ax^{10^b}\)) and how to use scientific notation effectively without losing precision, how to estimate ending digits in calculations.

We will discuss:

- fundamental units of the two measuring systems
- use of integers in scientific notation
- significant figures
- estimation of final digits
- precision, accuracy, and uncertainty

Assessment:

1. How many liters are in 2.64 pints?  
   1.2408 OR 1.2571 depending on which conversion factor you use

   4,790,000 ml

3. Scientific notation:  
   a. Approximate number of cells in the human body is 37,000,000,000,000 cells \(3.7 \times 10^{13}\)

   b. Approximate size of a human skin cell is 0.00003 meters \(3.0 \times 10^{-5}\) meters

4. Accuracy, Precision, and Uncertainty:  
   a. Precision: are the values 12.01 in, 12.00 in, 11.99 in, and 12.00 in precise?  
      YES, they are all close together

   b. Accuracy: are the values above accurate if the known length is 30 cm (11.81 in)?  
      NO (although in the real world, you’d have to do a statistical test and obtain a p-value to determine this)

5. Significant Figures, uncertain figures, and rounding:  
   a. 13.214 + 234.6 + 7.035 + 6.38 = 261.2; only 1 uncertain figure (#)

   b. 0.00435 x 4.6 = 0.020; least number of significant figures, in this case 2.

   c. 4503.67 + 34.90 x 5.724 = 4503.67 + 199.77 = 4703.44  
      Multiplication first, then round answer to least sig. figs. before doing addition. But, as a result of the exception to sig. figs. you can increase to 5 sig. figs. since the answer has a preceding 1, but there was not a preceding 1 in the original problem.

   d. 6.1625 x 2.00 = 12.32; don’t forget about the exception to rounding when dropping a 5 with no other non-zero digits, thus you round to the closes even digit.